

## WATER ACTIVITY CONCEPT FOR SAFETY FOOD STORAGE

### P. VULKOV

Bioevibul Ltd., 205, Alexander Stamboliisky Blvd., office 206, P.O. Box 51,

BG-1309 Sofia, Bulgaria

Tel./Fax: +359-2-8126563, e-mail: [office@bioevibul.com](mailto:office@bioevibul.com)

### ABSTRACT

It is well known that there are at least three factors of fundamental importance in determining the safety and longevity of stored food: water, temperature and oxygen. The outlines of this paper are water, especially the energy status of the water, measured and expressed in terms of the water activity. Water activity is still not commonly used, but has several distinct advantages for specifying conditions related to the safety of the foods and their storage. Basic principals and application of water activity in safety food storage are examined. On the base of concrete results for water activity of Bulgarian market foods and raw materials for their production, advantages of water activity management over measurement and controlling only the water content is demonstrated. The selected examples demonstrate that water activity measurement is quicker, easier and more informative than water content measurement.

### 1. INTRODUCTION

Two moisture measurements are required to completely characterize the state of water in a material such as a food product sample. The first measurement is water content. Water content is a measure of the amount of water in the sample and can be directly determined by drying a sample.

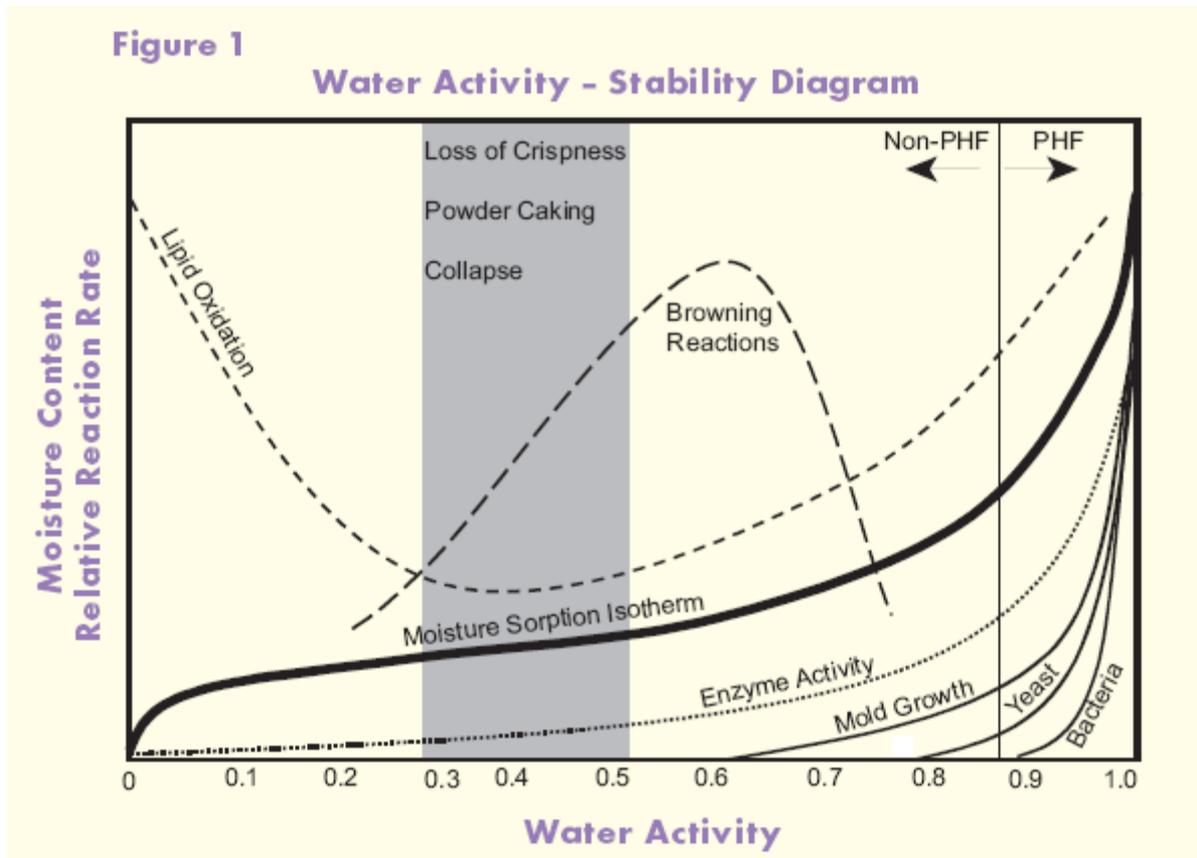
The second measurement required to characterize the state of water is the water activity. Water activity ( $a_w$ ) is the ratio of the vapour pressure of water in a material ( $p$ ) to the vapour pressure of pure water ( $p_0$ ) at the same temperature. Relative humidity of air is the ratio of the vapour pressure of air to its saturation vapour pressure. When vapour and temperature equilibrium are obtained, the water activity of the sample is equal to the relative humidity of air surrounding the sample in a sealed measurement chamber. Multiplication of water activity by 100 gives the equilibrium relative humidity (ERH) in percent.

$$a_w = p/p_0 = \text{ERH} (\%) / 100$$

As described by the above equation, water activity is a ratio of vapour pressures and thus has no units. It ranges from 0.0  $a_w$  (bone dry) to 1.0  $a_w$  (pure water). A water activity measurement of 0.80 means that the vapour pressure of the sample is eighty percent of the vapour pressure of pure water at the same conditions.

Water activity predicts safety and stability with respect to microbial growth, chemical and biochemical reaction rates, and physical properties. Figure 1 shows stability in terms of microbial growth limits and rates of degradative reactions as a function of water activity. Therefore, by measuring and controlling the water activity, it is possible to:

- predict which microorganisms will be potential sources of spoilage and infection;
- maintain the chemical stability of products;
- minimize nonenzymatic browning reactions and spontaneous autocatalytic lipid oxidation reactions;
- prolong the activity of enzymes and vitamins;
- optimize the physical properties of products such as moisture migration, texture, and shelf life.



## 2. EXPERIMENTAL DATA

The water activity ( $a_w$ ) was measured with a water activity meter AquaLab 3TE (Decagon, USA) - internal temperature control model, with thermoelectric (Peltier) components to maintain internal temperature. Though  $a_w$  of most products varies by less than  $\pm 0,002$  per  $^{\circ}\text{C}$ , the measurements was made at  $25^{\circ}\text{C}$ . AquaLab uses the chilled-mirror dew-point technique for measuring  $a_w$  of the sample. Because this is a primary measurement method, no calibration was necessary. The apparatus was only check (verified) for linear offset periodically by using a salt solution ( $a_w$  levels 0,984; 0,760; 0,500 and 0,250) and distilled water ( $a_w$  1,000).

Samples ware left intact if possible to ensure that the  $a_w$  is not altered. If necessary, the sample material was cut to fit the sampling cup. A sample that cannot be analysed immediately upon receipt was stored. For long term storage, glass jars or mylar bags was used. Once prepared, the sample was analyzed the same day. Refrigerated and frozen samples were prewarmed to ambient temperature to reduce the equilibration time and to give a more accurate reading. Multi-component samples or samples that have outside coatings was measured by crushing, slicing or grinding before putting it in the sample cup. In this case  $a_w$  was read as average water activity of the entire sample.

## 3. RESULTS AND DISCUSSION

### 3.1 Microbial Growth and Limiting Microorganism Growth

Water activity indicates the amount of water in the total water content which is available to micro-organisms. Each species of micro-organism (bacteria yeast and mould) has its own minimum  $a_w$  value below which growth is no longer possible. This limit corresponds to the 'suction power' of the various organisms, i.e. to the osmotic pressure they create and which must be higher than in the aqueous phase of food, to absorb from the food the water needed for metabolic activity and growth (see table below). By measuring the  $a_w$  value of foodstuffs it is possible to determine which micro-organisms will not be able to develop on them.

Typical growth limits function of  $a_w$  are 0.91 - 0.95 for most bacteria; 0.88 - most yeast; 0.80 - most mildew; 0.75 - halophile bacteria; 0.70 - osmiophil yeast; 0.65 - xerophile mildew. The water activity level that limits the growth of the vast majority of pathogenic bacteria is 0.90  $a_w$ , 0.70  $a_w$  for spoilage moulds, and the lower limit for all micro-organisms is 0.60  $a_w$ .

**Table 1 Foods with  $a_w$  range 1,00 – 0,95**

Microorganisms inhibited by lowest $a_w$ in range 1,00 – 0,95	<i>Pseudomonas, Escherichia, Proteus, Shigella, Klebsiella, Bacillus, Clostridium perfringens, some yeasts</i>		
Example of foods within this range			
Food	Ingredients	Water activity / temp. °C	Remarks
Mineral water		1,000 / 25°C	
Spread cheese	Cheese, curds, vegetable oil, emulsifier salts (E450, E452, E339, E1412), flavours	0,992 / 25°C	
Potato Balls	Potatoes, mashed potatoes, cottage cheese, breadcrumbs, onions, salt, sour cream, dill, parsley, black pepper, E 202, E 211	0,991 / 25°C	The classic vegetarian balls
Mayonnaise	Sunflower oil, vinegar, sugar, salt, E1414, E 330, E412, E415, , E202, E211,	0,989 / 25°C	
Bean dish (Gozba)	Beans, roasted and fried peppers, tomato puree, carrots, vegetable oil, tomatoes, sugar, aubergines, onions, salt, herbs, modified starch, E 202, E 211	0,989 / 25°C	Beans cooked from an authentic recipe
Hot Peppers in Tomato Sauce	Tomatoes, hot peppers, vegetable oil, salt, sugar, vinegar, modified starch, E 202, E 211	0,986 / 25°C	A serious Bulgarian lunch can't do without hot peppers
Plant production bread Dobrudja type	Wheat flour, yeast, salt	0,971 / 25°C	
Mustard	Mustard, refined sunflowers oil, vinegar, salt, sugar, spices, horse-radish, stabilizer (E412, E415, E224), and preservatives (E211, E330), water	0,970 / 25°C	
Bread Stara Zagora type cut in slices	Wheat flour, yeast, salt	0,964 / 25°C	
Tomato Chutney	Sweet peppers, tomato puree, carrots, vegetable oil, tomatoes, aubergines, sugar, salt, modified starch, E 202, E 211	0,955 / 20°C	The tomato chutney is one of the first industrially prepared canned foods in Bulgaria. Since the beginning, there is a great variety of brands of different companies.
"Snow-White" Salad	Strained yogurt, fresh cucumbers, mayonnaise, sour cream, dill, salt, garlic, E 202, E 211	0,950 / 20°C	"Snow-white" salad comes from Greece along with the Ouzo

**Table 2 Foods with  $a_w$  range 0,95 – 0,91**

Microorganisms inhibited by lowest $a_w$ in range 0,95 – 0,91	<i>Salmonella, V. parahaemolyticus, C. botulinum, Serratia, Lactobacillus, Pediococcus, some yeasts, yeasts (Rhodotorula, Pichia)</i>		
Example of foods within this range			
Food	Ingredients	Water activity /	Remarks

		temp. °C	
Ham from poultry fillet	Poultry fillet, salt, E250, E300, E330, E407, E452	0,942 / 25°C	
Caramel filling	oil, sugar, stabilizer, salt, vanillin, preservative	0,930 / 25°C	Filling for Croissant
Cheesy banitza	Wheat flour, cheese	0,929 / 25°C	savoury cheese pastry
White chocolate cream		0,920 / 20°C	

**Table 3 Foods with  $a_w$  range 0,91 – 0,87**

Microorganisms inhibited by lowest $a_w$ in range 0,91 – 0,87	Many yeasts ( <i>Candia</i> , <i>Torulopsis</i> , <i>Hansenula</i> ), <i>Micrococcus</i>		
Example of foods within this range			
Food	Ingredients	Water activity / temp. °C	Remarks
Dietetic Ham Sausage	Poultry meat, salt, E407, E250, E452, E621, E1442, E300	0,905 / 25°C	
Fermented sausage (salami)	veal meat, pork meat, bacon, salt, spices, E120, E250, E330	0,863 / 25°C	

**Table 4 Foods with  $a_w$  range 0,87 – 0,80**

Microorganisms inhibited by lowest $a_w$ in range 0,87 – 0,80	Most moulds (mycotoxigenic penicillia), <i>Staphylococcus aureus</i> , most <i>Sacharomyces</i> , <i>Debarymyces</i>		
Example of foods within this range			
Food	Ingredients	Water activity / temp. °C	Remarks
Raw-dried meat salami	veal, pork, salt and spices	0,847 / 25°C	
Jam "Strawberry"	strawberries, sugar, pectin E440, lemon acid E330	0,821 / 25°C	
Croissant with strawberry filling	Flour, strawberry jams filling, margarine, sugar, butter, egg powder, glucose, salt, and yeast. E471, E282	0,815 / 25°C	
Margarine	Vegetable oil, water, E322, E471, E330, E202, E160, E320, E321	0,810 / 25°C	
Jam "Raspberry"	raspberries, sugar, pectin E440, lemon acid E330	0,801 / 25°C	

**Table 5 Foods with  $a_w$  range 0,80 – 0,75**

Microorganisms inhibited by lowest $a_w$ in range 0,80 – 0,75	Most halophilic bacteria, mycotoxigenic aspergilli		
Example of foods within this range			
Food	Ingredients	Water activity / temp. °C	Remarks
Apricot jam	apricot, sugar, pectin E440, lemon acid E330	0,797 / 25°C	
Tomato Chutney Parvomaï	Sweet peppers puree, tomato puree, vegetable oil, sugar, salt, space	0,790 / 20°C	
Butter		0,780 / 25°C	

**Table 6 Foods with  $a_w$  range 0,75 – 0,65**

Microorganisms inhibited by lowest $a_w$ in range 0,75 – 0,65	Most xerophilic moulds ( <i>Aspergillus chevalieri</i> , <i>A. candidus</i> , <i>Wallemia sebi</i> ) <i>Sacharomyces bisporus</i>		
Example of foods within this range			
Food	Ingredients	Water activity / temp. °C	Remarks
Corn cereals		0,710 / 25°C	
Wheat cereals		0,675 / 25°C	
Wheat flour type 500		0,650 / 25°C	

**Table 7 Foods with  $a_w$  in range 0,65 – 0,61**

Microorganisms inhibited by lowest $a_w$ in range 0,80 – 0,75	Most osmophilic yeasts ( <i>Zygosacharomyces rouxii</i> ), few moulds ( <i>Aspergillus echinulatus</i> , <i>Monascus bisporus</i> )		
Example of foods within this range			
Food	Ingredients	Water activity / temp. °C	Remarks
Pasturma	Cured dried beef, mild flavour, spices, salt, vacuum packed.	0,640 / 25°C	
Wheat flour type 700		0,620 / 25°C	

**Table 8 Foods with  $a_w$  below 0,61**

Microorganisms inhibited by lowest $a_w$ below 0,61	No microbial proliferation		
Example of foods within this range			
Food	Ingredients	Water activity / temp. °C	Remarks
Wheat flour type 1150		0,570 / 25°C	
Dried plum		0,541 / 25°C	
Dried apple		0,503 / 25°C	
Finetti Cream brown		0,339 / 25°C	
Finetti Cream white		0,322 / 25°C	

### 3.2 Chemical Reactivity

Water activity influences not only microbial spoilage but also chemical and enzymatic reactivity. Water may influence chemical reactivity in different ways; it may act as a solvent, reactant, or change the mobility of the reactants by affecting the viscosity of the system. Water activity influences nonenzymatic browning, lipid oxidization, degradation of vitamins and other nutrients, enzymatic reactions, protein denaturation, starch gelatinization, and starch retrogradation (see Figure 1). Typically, as the water activity level is lowered, the rate of chemical degradative reactions decreases.

### 3.3 Physical Properties

Besides predicting the rates of various chemical and enzymatic reactions, water activity affects the textural properties of foods. Foods with high  $a_w$  (Table 1 and 2) have a texture that is described as moist, juicy, tender, and chewy. When the water activity of these products is lowered, undesirable textural attributes, such as hardness, dryness, staleness, and toughness, are observed. Low  $a_w$  products normally have texture attributes described as crisp and crunchy, while these products at higher  $a_w$  levels change to soggy texture. Critical water activities determine where products become unacceptable from a sensory standpoint.

### 3.4 Caking, Clumping, Collapse and Stickiness

Water activity is an important factor affecting the stability of powders and dehydrated products during storage. Controlling water activity in a powder product maintains proper product structure, texture, and stability, density, and dehydration properties. Knowledge of the water activity of powders as a function of moisture content and temperature is essential during processing, handling, packaging and storage to prevent the deleterious phenomenon of caking, clumping, collapse and stickiness. Caking is water activity, time, and temperature dependent and is related to the collapse phenomena of the powder under gravitational force.

### 3.5 Moisture Migration

Because water activity is a measure of the energy status of the water, differences in water activity between components is the driving force for moisture migration as the system comes to an equilibrium. Thus, water activity is an important parameter in controlling water migration of multicomponent products. Some foods contain components at different water activity levels, such as filled snacks or cereals with dried fruits. By definition, water activity dictates that moisture will migrate from a region of high  $a_w$  to a region of lower  $a_w$ , but the rate of migration depends on many factors. Undesirable textural changes can result from moisture migration in multicomponent foods. For example, moisture migrating from the higher  $a_w$  dried fruit into the lower  $a_w$  cereal causes the fruit to become hard and dry while the cereal becomes soggy.

#### 4. CONCLUSION

On the base of concrete results for water activity of Bulgarian market foods and raw materials for their production, advantages of water activity management over measurement and controlling only the water content is demonstrated. The selected examples demonstrate that water activity measurement is quicker, easier and more informative than water content measurement.

#### REFERENCES

- Fontana, A.J. Jr. (2000).** Water activity's role in food safety and quality, Decagon Devices, Inc. Pullman, Washington, USA. Presented at the Second NSF International Conference on Food Safety, October 11-13, 2000, Savannah, GA USA.
- Duckworth, R. (1975).** Water Relations of Foods. Academic Press, New York.
- Gomez-Diaz, R. (1992).** Water activity in foods: Determination methods. *Alimentaria*. 29:77-82.
- Prior, B.A. (1979).** Measurement of water activity in foods: A review. *Journal of Food Protection*. 42(8):668-674.
- Troller, J.A. and J.H.B. Christian. (1978).** Water Activity and Food. Academic Press, New York.
- Franks, F. (1982).** Water activity as a measure of biological viability and quality control. *Cereal Foods World*. 27(9):403-407.
- Hardman, T.M. (1988).** Water and Food Quality. Elsevier Press, London.
- Rockland, L.B. and G.F. Stewart. (1981).** Water Activity: Influences on Food Quality. Academic Press, New York.
- Beuchat, L.R. (1981).** Microbial stability as affected by water activity. *Cereal Foods World*. 26(7):345-349.
- Farber, J.M., F. Coates, and E. Daley. (1992).** Minimum water activity requirements for the growth of *Listeria monocytogenes*. *Letters In Applied Microbiology*. 15:103-105.
- Kuntz, L.A. (1992).** Keeping microorganisms in control. *Food Product Design*. August:44-51.
- Tokuoka, K. and T. Ishitani. (1991).** Minimum water activities for the growth of yeasts isolated from high-sugar foods. *Journal of General and Applied Microbiology*. 37:111-119.
- Lee, M.B. and S. Styliadis. (1996).** A survey of pH and water activity levels in processed salamis and sausages in Metro Toronto. *Journal of Food Protection*. 59:1007-1010.
- Pisecky, J. (1992).** Water activity of milk powders. *Milchwissenschaft*. 47:3-7.